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Key indicators

Single-crystal X-ray study

$T = 293$ K

Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å

R factor = 0.048

wR factor = 0.155

Data-to-parameter ratio = 14.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

3-Benzotriazol-1-yl 5-*tert*-butyl 2,6-dimethyl-4-(3-nitrophenyl)-1,4-dihydropyridine-3,5-dicarboxylate

The title compound, $\text{C}_{25}\text{H}_{25}\text{N}_5\text{O}_6$, is an important intermediate in the synthesis of nefidipine-type pharmaceuticals. The crystal packing is stabilized by intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds.

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Comment

4-Aryl-1,4-dihydropyridine-3,5-dicarboxylic diesters of the nefidipine type have become almost indispensable for the treatment of cardiovascular diseases since they first appeared on the market in 1975 (Yiu & Knaus, 1999; Goldmann & Stoltefuss, 1991). The title compound, (I), is a key intermediate for their preparation.

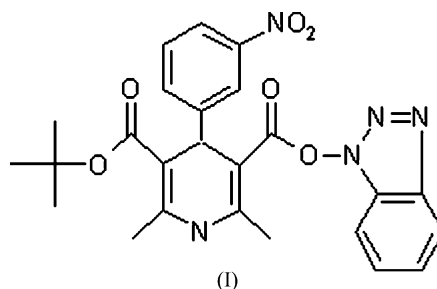


Fig. 1 shows the structure of the title compound. The molecule contains an aromatic ring, $R1$ (C13–C18), a dihydropyridine ring, $R2$, and a benzotriazole ring system, $R3$. The dihedral angles for $R1/R2$, $R1/R3$ and $R2/R3$ are $88.3(2)$, $43.4(2)$ and $92.3(2)^\circ$, respectively. This compares well with

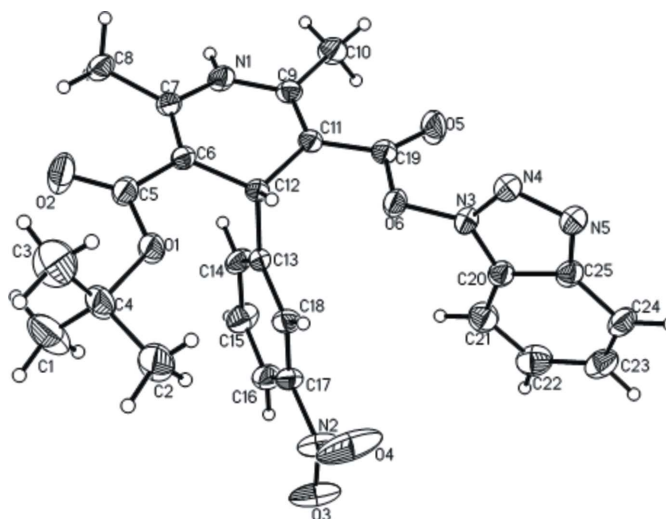


Figure 1

A view of the title compound. Displacement ellipsoids are drawn at the 30% probability level.

the values for nefidipine (Hofmann & Cimraglia, 1990; Ramusino & Vari, 1999).

An intermolecular N—H...O hydrogen bond links the molecules into infinite chains (Table 1).

Experimental

2,6-Dimethyl-4-(3-nitro-phenyl)-1,4-dihydropyridine-3,5-dicarboxylic acid mono-*tert*-butyl ester (491 mg, 1 mmol) was dissolved in CH₂Cl₂ (30 ml); dicyclohexylcarbodiimide (206 mg, 1 mmol) and benzotriazol-1-ol (135 mg, 1 mmol) in CH₂Cl₂ (10 ml) were added to the solution at 278 K. The reaction mixture was stirred at 276–279 K for a further 10 h. The solvent CH₂Cl₂ was removed by vacuum evaporation at 293 K. The product was purified by chromatography on a silica gel column (eluted by ethyl acetate and petroleum ether, 1:5) at room temperature with a yield of 92% (450 mg). Suitable crystals were obtained by slow evaporation of a solution in methanol.

Crystal data

C ₂₅ H ₂₅ N ₅ O ₆	$D_x = 1.302 \text{ Mg m}^{-3}$
$M_r = 491.50$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 19215 reflections
$a = 10.332 (2) \text{ \AA}$	$\theta = 3.3\text{--}25.5^\circ$
$b = 15.163 (3) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$c = 16.010 (3) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 90.96 (3)^\circ$	Rod, yellow
$V = 2507.6 (9) \text{ \AA}^3$	$0.38 \times 0.25 \times 0.11 \text{ mm}$
$Z = 4$	

Data collection

Rigaku R-AXIS RAPID IP area-detector diffractometer oscillation scans	4620 independent reflections
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)	3165 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.965$, $T_{\max} = 0.989$	$R_{\text{int}} = 0.050$
23558 measured reflections	$\theta_{\text{max}} = 25.5^\circ$
	$h = -12 \rightarrow 12$
	$k = -18 \rightarrow 18$
	$l = -19 \rightarrow 19$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.048$	$w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$
$wR(F^2) = 0.155$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.01$	$(\Delta/\sigma)_{\text{max}} < 0.001$
4620 reflections	$\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$
325 parameters	$\Delta\rho_{\text{min}} = -0.23 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
$\text{N1--H1D}\cdots\text{O4}^i$	0.86	2.49	3.265 (3)	151

Symmetry code: (i) $x - 1, y, z$.

H atoms were placed in calculated positions and constrained to ride on their parent atoms, with C—H = 0.93–0.98 \AA , N—H = 0.86 \AA and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C,N})$ and $1.5U_{\text{eq}}(\text{methyl C})$.

Data collection: *RAPID-AUTO* (Rigaku, 2004); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

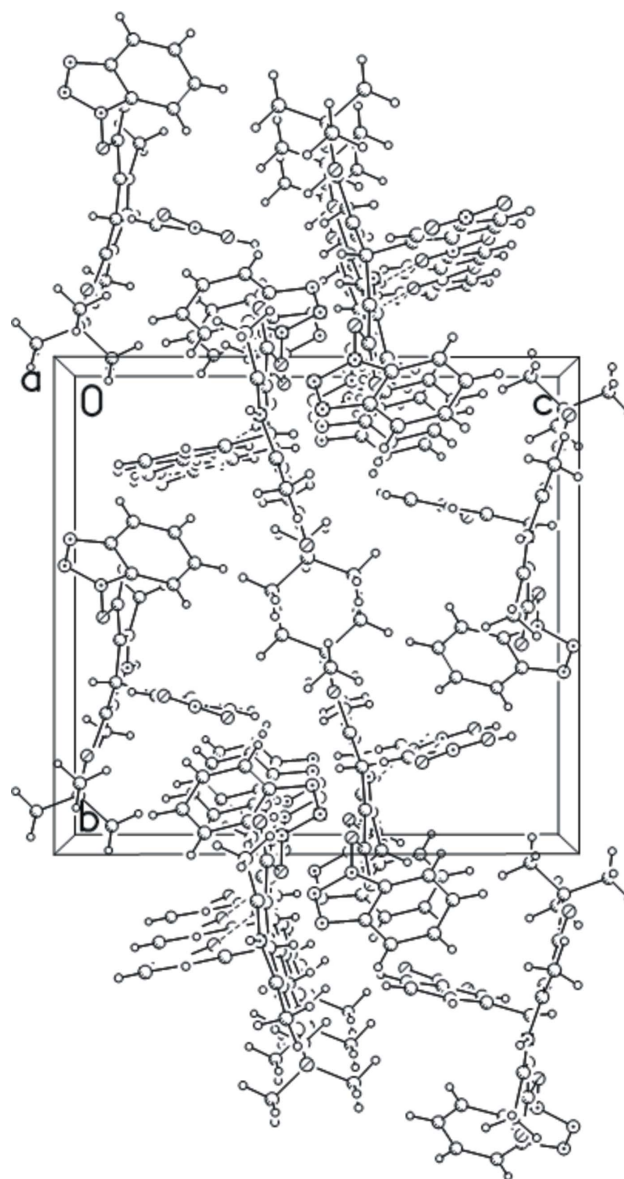


Figure 2
The packing of (I).

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